Adding stable carbon isotopes improves model representation of the role of microbial communities in peatland methane cycling

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Adding stable carbon isotopes improves model representation of the role of microbial communities in peatland methane cycling

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Abstract

Climate change is expected to have significant and uncertain impacts on methane (CH\textsubscript{4}) emissions from northern peatlands. Biogeochemical models can extrapolate site-specific CH\textsubscript{4} measurements to larger scales and predict responses of CH\textsubscript{4} emissions to environmental changes. However, these models involve considerable uncertainties and limitations in representing CH\textsubscript{4} production, consumption, and transport processes. To improve predictions of CH\textsubscript{4} transformations, we incorporated acetate and stable carbon (C) isotopic dynamics associated with CH\textsubscript{4} cycling into a biogeochemistry model, DNDC. By including these new features, DNDC explicitly simulates acetate dynamics and the relative contribution of acetotrophic and hydrogenotrophic methanogenesis (AM and HM) to CH\textsubscript{4} production, and predicts the C isotopic signature ($\delta^{13}C$) in soil C pools and emitted gases. When tested against biogeochemical and microbial community observations at two sites in a zone of thawing permafrost in a subarctic peatland in Sweden, the new formulation substantially improved agreement with CH\textsubscript{4} production pathways and $\delta^{13}C$ in emitted CH\textsubscript{4} ($\delta^{13}C$-CH\textsubscript{4}), a measure of the integrated effects of microbial production and consumption, and of physical transport. We also investigated the sensitivity of simulated $\delta^{13}C$-CH\textsubscript{4} to C isotopic composition of substrates and, to fractionation factors for CH\textsubscript{4} production ($\gamma_{\text{AM}}$ and $\gamma_{\text{HM}}$), CH\textsubscript{4} oxidation ($\gamma_{\text{Ox}}$), and plant-mediated CH\textsubscript{4} transport ($\gamma_{\text{TP}}$). The sensitivity analysis indicated that the $\delta^{13}C$-CH\textsubscript{4} is highly sensitive to the factors associated with microbial metabolism ($\gamma_{\text{AM}}$, $\gamma_{\text{HM}}$, and $\gamma_{\text{Ox}}$). The model framework simulating stable C isotopic dynamics provides a robust basis for better constraining and testing microbial mechanisms in predicting CH\textsubscript{4} cycling in peatlands.

1. Introduction

Northern peatlands are characterized by cold, wet conditions that promote the sequestration of atmospheric carbon dioxide (CO\textsubscript{2}) into surface soil organic carbon (SOC) or peat [e.g., Gorham, 1991; Zimov et al., 2006; Schuur et al., 2008]. These peatlands have accumulated 473–621 Pg (10\textsuperscript{15} g) carbon (C) since the Last Glacial Maximum [Yu et al., 2010], with more than 277 Pg C stored in permafrost areas [Schuur et al., 2008; Tarnocai et al., 2009]. Although northern peatlands are currently a net C sink, they are an important source of atmospheric methane (CH\textsubscript{4}), releasing 31–65 Tg (10\textsuperscript{12} g) CH\textsubscript{4} yr\textsuperscript{-1} [McGuire et al., 2009]. Recent studies indicate that the rate and extent of permafrost degradation is increasing with pronounced climate warming in northern peatlands [e.g., James et al., 2013; Payette et al., 2004; Quinton et al., 2011; Åkerman and Johansson, 2008]. Permafrost thaw can result in changes in topography (e.g., thermokarst), soil climate, and vegetation [e.g., Avis et al., 2011; Malmer et al., 2005; Schuur et al., 2008]. Changes associated with climate warming and permafrost degradation mobilizes previously frozen C, and stimulates microbial decomposition of peat C stocks into the climate forcing trace gases CO\textsubscript{2} and CH\textsubscript{4} [e.g., Dorrepaal et al., 2009; Frolking et al., 2011; Johnston et al., 2014; McGuire et al., 2009; Schuur et al., 2009, 2011]. Increased emissions of C gases, CH\textsubscript{4} in...
particular due to its strong global warming potential (28 kg CO₂-equivalents kg⁻¹ CH₄ at a 100 year time-scale) [IPCC, 2013], would increase the rate of warming, forming a significant positive biogeochemical feedback to climate [e.g., Olefeldt et al., 2013].

Despite considerable research attention focused on CH₄ emissions from northern peatlands, large uncertainty over their magnitude and variability remains [e.g., Limpens et al., 2008; McGuire et al., 2009; Olefeldt et al., 2013]. Northern peatlands are highly heterogeneous, usually with varying characteristics of permafrost, topography, hydrology, soil, and vegetation within close proximity [Eppinga et al., 2009], which results in considerable variations of CH₄ fluxes at local and landscape scales [e.g., Bäckstrand et al., 2010; Lund et al., 2010; Sachs et al., 2010]. To extrapolate site-specific CH₄ measurements to larger regions and/or predict responses of CH₄ emissions to environmental changes, process-based biogeochemical models have been developed and applied from site to global scales [e.g., Wania et al., 2013; Melton et al., 2013; Xu et al., 2016].

While many improvements have been made in modeling CH₄ emissions from northern areas by incorporating thermal, hydrologic, vegetation, and biogeochemical processes in relation to permafrost characteristics into model frameworks [e.g., Schneider von Diemling et al., 2012; Wania et al., 2009a, 2009b; Zhuang et al., 2001, 2004], key limitations and uncertainties still exist. There are a number of important controls over CH₄ production, consumption, and transport that have not been, or are inadequately, incorporated into existing CH₄ biogeochemical models [Bridgham et al., 2013].

For example, most existing models use net primary production or SOC as an index to represent substrate availability for CH₄ production and do not differentiate CH₄ production pathways, i.e., acetotrophic methanogenesis (AM) and hydrogenotrophic methanogenesis (HM), although mechanisms of CH₄ production are usually addressed differently in the models [Bridgham et al., 2013; Melton et al., 2013; Xu et al., 2016]. In addition, the ability to test model simulations of the individual processes of CH₄ production, oxidation, and transport is still limited because only a few model parameters and/or processes can be constrained by comparing models against measurements of net CH₄ emissions [Bridgham et al., 2013].

The process-based biogeochemical model, DeNitrification-DeComposition (DNDC), incorporates both the AM and HM pathways for CH₄ production [Fumoto et al., 2008], and has a permafrost thermal submodel, based on the Northern Ecosystem Soil Temperature (NEST) model, in order to simulate high-latitude soil biogeochemistry in permafrost ecosystems, and has been tested against CH₄ flux data measured at several northern peatlands [Deng et al., 2014, 2015; Zhang et al., 2012]. However, new data characterizing the methanogen community at Stordalen Mire in Sweden indicate that HM lineages dominate the methanogen community in early thaw stages [McCalley et al., 2014; Mondav et al., 2014] whereas DNDC simulations show primarily AM. Given the importance of this microbially mediated process in regulating CH₄ dynamics across the permafrost thaw gradient [McCalley et al., 2014], this discrepancy in simulating CH₄ production pathway may hinder reliable prediction of responses of CH₄ emissions to permafrost degradation. Since different CH₄-cycling microbes fractionate ¹³C differently [Conrad, 2005], adding stable C isotopic information to soil organic C processing in DNDC is a potentially powerful tool for generating predictions, based on microbial CH₄-cycling mechanisms, that can be tested against observations of the ¹³C isotopic composition of emitted gases. High temporal frequency measurements of the magnitude and isotopic composition of CH₄ emissions have been made at Stordalen Mire on plots with and without permafrost [McCalley et al., 2014], providing an excellent case study.

To address both the simulation discrepancy of AM versus HM production and uncertainty in the description of the net flux versus individual processes of CH₄ production, oxidation, and transport, we made two significant modifications to DNDC simulations of CH₄ cycling. First, stable C isotopic values of soil C pools and fractionation impacts of C processing pathways were added to the model. Second, the anaerobic fermentation pathway in DNDC was modified to include the production of an intermediate pool—acetate—which can be consumed in AM or by oxidation by other terminal electron acceptors. In this study, we report the details of these modifications, a sensitivity assessment, and evaluation of the new model formulation against field data from Stordalen Mire.

### 2. The Study Area and Field Data

Field data used for this study were collected at a subarctic peatland with discontinuous, ice-rich permafrost, Stordalen Mire, (68°20’N, 19°03’E, 351 m.a.s.l.) located 10 km southeast of the Abisko Scientific Research
As in most peatlands in discontinuous permafrost regions, Stordalen Mire has high spatial heterogeneity in topography (1–2 m relative differences in elevation). The variability of topography creates spatially restricted environments (on the scale of m$^2$) with different soil moisture and nutrient conditions that support different plant communities [Roswall et al., 1975; Bäckstrand et al., 2008]. Measurements at Stordalen Mire have been made at three sites that represent three dominant cover types, Palsa, Sphagnum, and Eriophorum (note that in this study the terms Sphagnum and Eriophorum are used to denote land cover types instead of vegetation species, though the site terms are derived from the dominant vegetation). The Palsa sites are dry features underlain by permafrost, with an active layer thickness (ALT; the thickness of the surface soil layer that freezes and thaws seasonally above a year-round frozen layer) usually <0.7 m in late summer. The Sphagnum sites are partially underlain by permafrost, representing intermediate thaw features, with an ALT generally thicker than 1.0 m in late summer and water table levels fluctuating within 0–25 cm below the ground surface. The plant community structure, the fluctuating water table, and the pore water chemistry indicate that they are ombrotrophic, perched above the local water table. The Eriophorum sites have no detectable permafrost and are generally wetter than Sphagnum, with water table levels generally at or above the peat surface [Bäckstrand et al., 2008, 2010; Olesen et al., 2012], and are minerotrophic. During the last several decades, there have been pronounced shifts in the extent of these three land cover types, with some drier sites getting wetter and shifting vegetation cover as permafrost thaws [Christensen et al., 2004; Malmer et al., 2005]. These three land cover types can be regarded as representing a natural landscape gradient of permafrost thaw [e.g., Malmer et al., 2005; Johansson et al., 2006; Bäckstrand et al., 2010; Hodgkins et al., 2014, 2015].

There are multiyear field records of CO$_2$ and CH$_4$ fluxes at Stordalen, measured using automated chambers. In a previous study, we validated the DNDC model against the multiyear (2003–2009) field records, including soil freeze/thaw dynamics, NEE, and CH$_4$ fluxes, and the validation indicated that DNDC was able to simulate the observed differences in seasonal soil thaw, NEE, and CH$_4$ fluxes across the three cover types [Deng et al., 2014].

Since 2011, additional field studies have been conducted at Stordalen with a focus on measuring the role of microbial communities in regulating CH$_4$ dynamics [Hodgkins et al., 2014; McCalley et al., 2014; Mondav et al., 2014]. Daily averaged values of CH$_4$ flux rates and $\delta^{13}$C of emitted CH$_4$ were derived from frequent (subdaily) measurements made at the Sphagnum and Eriophorum sites from 2011 to 2013 using a Quantum Cascade Laser Spectrometer (QCLS, Aerodyne Research Inc.) connected to an automated chamber system [McCalley et al., 2014]. The relative abundance of methanogens was also quantified in peat collected near the auto chambers four times through the 2011 growing season (15 June, 12 July, 15 August, and 15 October) using 16S rRNA gene amplicon sequencing (described in detail in McCalley et al. [2014] and Mondav et al.[2014]). In addition, soil water table depth (WTD, positive values up from the ground surface) was measured three to five times per week from June to October each year [McCalley et al., 2014]. Daily meteorological data, including air temperature, precipitation, solar radiation, wind speed, and relative humidity, were recorded at ANS (Figure 1).

The field data indicated that: (1) CH$_4$ fluxes and isotopic composition were significantly different between the study sites, with an increasing trend for both CH$_4$ fluxes and $\delta^{13}$C-CH$_4$ with permafrost thaw, (2) CH$_4$ production pathways were different between the sites and a distinct shift from HM to AM was observed along the thaw gradient, and (3) microbial community and CH$_4$ production pathway played an important role in regulating CH$_4$ dynamics at Stordalen Mire, and could be crucial for improving model predictions of CH$_4$ feedbacks in response to climate change and permafrost thaw [McCalley et al., 2014; Mondav et al., 2014]. Driven by the discrepancies between the simulated and observed relative contributions of different...
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3. The DNDC Model

DNDC is a process-based model developed to quantify C sequestration as well as the emissions of C and nitrogen (N) gases from terrestrial ecosystems [Li et al., 1992a, 1992b, 2000]. The model has incorporated a relatively large suite of biophysical and biogeochemical processes to compute the complex transport and transformations of C and N in terrestrial ecosystems under both aerobic and anaerobic conditions. DNDC is comprised of three major interacting submodels: soil climate, plant growth, and soil biogeochemistry. The soil climate and plant growth submodels convert the primary drivers, such as climate, soil properties, vegetation, and anthropogenic activity, into soil environmental factors, such as soil temperature and moisture, pH, redox potential (Eh), and substrate concentrations. The soil biogeochemistry submodel, including processes of decomposition, nitrification, denitrification, and fermentation, simulate C and N transformations that are mediated by soil microbes and controlled by soil environmental factors [Li, 2000; Li et al., 2012].

In DNDC, the rate of CH₄ emission is predicted by modeling its production, consumption, and transport processes. Production is simulated by calculating substrate concentrations (i.e., electron donors and acceptors) resulting from decomposition of SOC as well as plant root activities including exudation and respiration, and then by tracking a series of reductive reactions between electron donors (i.e., H₂ and dissolved organic carbon (DOC)) and acceptors (i.e., NO₃, Mn⁴⁺, Fe³⁺, SO₂⁻, and CO₂). SOC decomposition is simulated by disaggregating SOC into four pools (i.e., litter, microbes, humads, and passive humus), and each pool is further divided into two or three subpools with specific C:N ratios and decomposition rates. During decomposition of each SOC subpool, the model distributes a fixed fraction of the released C into DOC divided into two or three subpools with specific C:N ratios and decomposition rates. During decomposition of each SOC subpool, the model distributes a fixed fraction of the released C into DOC and C:N ratios for each subpool are determined by Eh calculations [Li et al., 1992a, 1992b, 2000]. When a soil is shifting from unsaturated to saturated conditions, soil oxygen is gradually depleted and additional oxidants (e.g., NO₃, Mn⁴⁺, Fe³⁺, SO₂⁻, and CO₂) may become involved in reductive reactions. Soil Eh decreases along with the consumption of these oxidants and DNDC simulates denitrification, reductions of Mn⁴⁺, Fe³⁺, and SO₂⁻, and methane production as consecutive reactions with each reaction occurring under certain Eh conditions [Li et al., 2004]. The model simulates sequential reactions from NO₃ to N₂ in denitrification and calculates the rate of each step based on the concentrations of the corresponding nitrobenzoic oxides and DOC [Li et al., 1992a]. Reductions of Mn⁴⁺, Fe³⁺, and SO₂⁻ were simulated using dual-substrate Michaelis-Menten-based equations. Maximum rates for Mn⁴⁺, Fe³⁺, and SO₂⁻ reductions are set as 0.108 mol kg⁻¹ d⁻¹, 0.108 mol kg⁻¹ d⁻¹, and 0.691 mol m⁻³ d⁻¹, respectively. Michaelis-Menten half-saturation constants for Mn⁴⁺, Fe³⁺, and SO₂⁻ are 0.15 mol kg⁻¹, 0.15 mol kg⁻¹, and 0.23 mol m⁻³, respectively [Fumoto et al., 2008]. Michaelis-Menten half-saturation constants for DOC and H₂ are 0.46 mol m⁻³ and 0.22 mmol m⁻³, respectively, for the Mn⁴⁺ or Fe³⁺ reduction, and are 1.6 mol m⁻³ and 2.87 mmol m⁻³, respectively, for the SO₂⁻ reduction [Fumoto et al., 2008]. DNDC simulates methane production after depletions of NO₃, Mn⁴⁺, Fe³⁺, and SO₂⁻, when soil Eh is below −150 mV [Li et al., 2004]. Methane consumption is simulated as an oxidation reaction involving electron exchange between CH₄ and oxygen. In DNDC, CH₄ production and oxidation can occur simultaneously within a soil layer (typically 2–5 cm thick) but within relatively aerobic and anaerobic sublayers, whose volumetric fractions are determined by Eh calculations [Li, 2007].
Redox potential, temperature, pH, and the concentrations of electron donors and acceptors are the major factors controlling the rates of CH₄ production and oxidation. Methane transport from soil into atmosphere is simulated via three ways, including plant-mediated transport, ebullition, and diffusion [Fumoto et al., 2008; Zhang et al., 2002].

4. Modifications of DNDC

In this study, we modified the version of DNDC used for simulating high-latitude soil biogeochemistry [Deng et al., 2014; Zhang et al., 2012]. We hypothesized that the DNDC’s capacity for simulating the relative contribution of each methanogenic pathway to total CH₄ production could be improved by explicitly simulating acetate dynamics, and then using acetate (instead of bulk DOC) as a substrate for acetoclastic methanogenesis. Incorporating stable C isotopic dynamics in CH₄ transformations could provide a method for testing and constraining the mechanisms of CH₄ cycling (production, oxidation, and transport).

4.1. Modeling Acetate Dynamics and CH₄ Production

To explicitly simulate acetate dynamics and CH₄ production pathways of AM and HM, we have introduced new processes to simulate acetate production and consumption, which jointly determine the concentration of acetate in peat soils (Figure 2). In the modified DNDC, acetate and H₂ are the immediate electron donors for AM and HM, respectively, in anaerobic soils, and the rates of the reactions are controlled by the availability of acetate and H₂ (Figure 2). These substrates can be produced through the mineralization of soil organic matter and anaerobic decomposition of complex dissolved organic substances. Based on Conrad [1999] and Van Bodegom and Scholten [2001], this process can be summarized as:

Figure 2. The framework for simulating soil biogeochemistry and methane dynamics in the modified DNDC. The model predicts the rate of CH₄ emission by modeling CH₄ production, consumption, and transport processes. CH₄ production is simulated by calculating substrate concentrations (i.e., electron donors and acceptors) resulting from decomposition of SOC as well as root exudation and respiration, and then by tracking a series of reductive reactions between electron donors (i.e., H₂ and acetate) and acceptors (i.e., NO₃⁻, Mn⁴⁺, Fe³⁺, SO₄²⁻, and CO₂) using Michaelis-Menten-based equations [Li et al., 2004; Fumoto et al., 2008]. CH₄ consumption is simulated as an oxidation reaction. CH₄ transport from soil into atmosphere is simulated via three ways, plant-mediated transport, ebullition, and diffusion [Fumoto et al., 2008; Zhang et al., 2002].
C₆H₁₂O₆ + 2H₂O → 2CH₃COOH + 2CO₂ + 4H₂  \quad (R1)

In DNDC, C₆H₁₂O₆ is represented by DOC resulting from both mineralization of SOC and plant root exudation. The reaction rate is calculated by using the Michaelis-Menten equation as follows:

\[ V_{\text{AnDecom}} = V_{\text{Max,DOC}} \cdot \frac{[\text{DOC}]}{K_{\text{AnDecom}} + [\text{DOC}]} \quad (E1) \]

where, \( V_{\text{Max,DOC}} \) is the maximum rate of anaerobic DOC decomposition, when DOC concentration is not limiting, \([\text{DOC}]\) is the concentration of DOC (mol m\(^{-3}\)), and \( K_{\text{AnDecom}} \) is the Michaelis half-saturation constant for this reaction. \( V_{\text{Max,DOC}} \) is calculated by using the maximum rate of anaerobic DOC decomposition at a reference temperature (20°C for this reaction) and a Q\(_{10}\) value of 2.0 for the temperature sensitivity.

Based on the stoichiometry in R1, the consumption of DOC and production of acetate, H₂, and CO₂ (note that CO₂ is also produced by other pathways, such as plant root respiration, AM, CH₄ oxidation, and among others, in DNDC) resulting from anaerobic DOC decomposition are calculated as follows:

\[ V_{\text{AnDecom}} = -\frac{d[\text{DOC}]}{dt} = 0.5 \cdot \frac{d[\text{CH₃COOH}]}{dt} = 0.5 \cdot \frac{d[\text{CO₂}]}{dt} = 0.25 \cdot \frac{d[H₂]}{dt} \quad (E2) \]

Methane is primarily produced from acetotrophic and hydrogenotrophic methanogenesis for most environments, with acetate and H₂/CO₂ being substrates of these two pathways, respectively [Conrad, 1989]. The reactions of acetotrophic and hydrogenotrophic methanogenesis are the following [Conrad, 1989]:

CH₃COOH → CO₂ + CH₄ \quad (R2)

CO₂ + 4H₂ → CH₄ + 2H₂O \quad (R3)

Michaelis-Menten kinetics are used for the reaction rates of the R2 and R3 [Van Bodegom and Scholten, 2001] as follows:

\[ V_{\text{AM}} = V_{\text{AM,Max}} \cdot \frac{[\text{CH₃COOH}]}{K_{\text{AM}} + [\text{CH₃COOH}]} \quad (E3) \]

\[ V_{\text{HM}} = V_{\text{HM,Max}} \cdot \frac{[H₂]}{K_{\text{HM}} + [H₂]} \quad (E4) \]

where, \( V_{\text{AM,Max}} \) and \( V_{\text{HM,Max}} \) are the maximum rates of acetotrophic and hydrogenotrophic methanogenesis, respectively, when concentrations of substrates are not limiting, [CH₃COOH] and are concentrations of acetate (mol m\(^{-3}\)) and H₂ (mol m\(^{-3}\)), respectively, and \( K_{\text{AM}} \) and \( K_{\text{HM}} \) (Table 1) are the corresponding Michaelis half-saturation constants for these two reactions. Both \( V_{\text{AM,Max}} \) and \( V_{\text{HM,Max}} \) are sensitive to temperature with a Q\(_{10}\) value of 4.6 [Van Bodegom and Scholten, 2001]. Total CH₄ production rate is calculated as the sum of the rates of acetate-dependent and H₂/CO₂-dependent methanogenesis.

The processes of CH₄ production consume acetate, H₂, and CO₂, and produce CO₂ (R2 and R3). Based on the stoichiometry in CH₄ production, the consumption of acetate and production of CO₂ in R2 are calculated as:

\[ V_{\text{AM}} = -\frac{d[\text{CH₃COOH}]}{dt} = \frac{d[\text{CO₂}]}{dt} \quad (E5) \]

The consumptions of CO₂ and H₂ in R3 are calculated as:

\[ V_{\text{HM}} = -0.25 \frac{d[H₂]}{dt} = -\frac{d[\text{CO₂}]}{dt} \quad (E6) \]

In addition to acetate consumption due to AM (R2), it has been widely observed that acetate can be consumed by other pathways, including oxidation by O₂ under aerobic conditions and electron transfer between acetate and humic substances (ETAH) in anaerobic peat soils [e.g., Duddleston et al., 2002; Lovley et al., 1996; Segers and Kengen, 1998]. Following Segers and Kengen [1998], DNDC simulates acetate consumption due to oxidation and ETAH based on the Michaelis-Menten equation:
Table 1. The Model Parameters for Simulating Acetate Dynamics, Acetotrophic, Methanogenesis (AM) and Hydrogenotrophic Methanogenesis (HM), and Stable C Isotopic Dynamics at the Sphagnum and Eriophorum Sites

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Description</th>
<th>Sphagnum</th>
<th>Eriophorum</th>
<th>Sources</th>
</tr>
</thead>
<tbody>
<tr>
<td>( V_{\text{Max, DOC}} )</td>
<td>Maximum anaerobic DOC decomposition rate, mol m(^{-3}) d(^{-1})</td>
<td>0.5 • DOC</td>
<td>0.5 • DOC</td>
<td>Calibrated</td>
</tr>
<tr>
<td>( K_{\text{HIDecom}} )</td>
<td>Michaelis half-saturation constant for anaerobic DOC decomposition, mol m(^{-3})</td>
<td>1.0</td>
<td>1.0</td>
<td>Calibrated</td>
</tr>
<tr>
<td>( V_{\text{AM, Max}} )</td>
<td>Maximum rate for AM, mmol kg(^{-1}) dry peat d(^{-1})</td>
<td>0.8</td>
<td>4.0</td>
<td>Calibrated</td>
</tr>
<tr>
<td>( KA )</td>
<td>Michaelis half-saturation constants for AM, mol m(^{-3})</td>
<td>2.56</td>
<td>2.56</td>
<td>(1)</td>
</tr>
<tr>
<td>( V_{\text{HM, Max}} )</td>
<td>Maximum rate for HM, mmol kg(^{-1}) dry peat d(^{-1})</td>
<td>7.2</td>
<td>4.0</td>
<td>Calibrated</td>
</tr>
<tr>
<td>( K_{\text{ATEA}} )</td>
<td>Michaelis half-saturation constants for HM, mol m(^{-3})</td>
<td>0.0133</td>
<td>0.0133</td>
<td>(1)</td>
</tr>
<tr>
<td>( V_{\text{CA, Eriophorum, Max}} )</td>
<td>Maximum rate for acetate consumption due to nonmethanogenesis pathways, mmol electron kg(^{-1}) dry peat s(^{-1})</td>
<td>10 (^{-2})</td>
<td>10 (^{-2})</td>
<td>(2)</td>
</tr>
<tr>
<td>( K_{\text{CA, Sphagnum}} )</td>
<td>Michaelis half-saturation constants of acetate for acetate consumption due to nonmethanogenesis pathways, mol m(^{-3})</td>
<td>0.01</td>
<td>0.01</td>
<td>(2)</td>
</tr>
<tr>
<td>( K_{\text{TE}, Sphagnum}} )</td>
<td>Michaelis half-saturation constants of electron acceptors for acetate consumption due to nonmethanogenesis pathways, mol m(^{-3})</td>
<td>10</td>
<td>10</td>
<td>(2)</td>
</tr>
<tr>
<td>( a )</td>
<td>Coefficient quantifying the concentration of the electron acceptors provided by humic substances, mmol electron g(^{-1}) C</td>
<td>2.0</td>
<td>1.5</td>
<td>Calibrated</td>
</tr>
<tr>
<td>( \delta^{13}C_{\text{DOC}} )</td>
<td>C isotope signature of DOC</td>
<td>−26(^{\text{‰}})</td>
<td>−26(^{\text{‰}})</td>
<td>(3)</td>
</tr>
<tr>
<td>( \delta^{13}C_{\text{CO}_{2}} )</td>
<td>C isotope signature for CO(_2) from root respiration</td>
<td>−26(^{\text{‰}})</td>
<td>−26(^{\text{‰}})</td>
<td>(3)</td>
</tr>
<tr>
<td>( \alpha_{D} )</td>
<td>Fractionation factor for DOC decomposition</td>
<td>1.000</td>
<td>1.000</td>
<td>(4)</td>
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<tr>
<td>( \alpha_{AM} )</td>
<td>Fractionation factor for AM</td>
<td>1.026</td>
<td>1.026</td>
<td>Calibrated</td>
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<tr>
<td>( \alpha_{HM} )</td>
<td>Fractionation factor for HM</td>
<td>1.073</td>
<td>1.073</td>
<td>Calibrated</td>
</tr>
<tr>
<td>( \alpha_{\text{TE}} )</td>
<td>Fractionation factor for CH(_2) oxidation</td>
<td>1.025</td>
<td>1.025</td>
<td>Calibrated</td>
</tr>
<tr>
<td>( \gamma_{\text{TE}} )</td>
<td>Fractionation factor for CH(_2) emission via plant-mediated transport</td>
<td>1.016</td>
<td>1.016</td>
<td>Calibrated</td>
</tr>
<tr>
<td>( \gamma_{\text{TE}} )</td>
<td>Fractionation factor for CH(_2) emission via ebullition</td>
<td>1.000</td>
<td>1.000</td>
<td>J. Chanton (personal communication, 2015)</td>
</tr>
<tr>
<td>( \delta_{\text{TD}} )</td>
<td>Fractionation factor for CH(_4) emission due to diffusion</td>
<td>1.001</td>
<td>1.001</td>
<td>J. Chanton (personal communication, 2015)</td>
</tr>
</tbody>
</table>

\[ V_{\text{CA, Eriophorum, Max}} = 0.125V_{\text{CA, Sphagnum, Max}} = \frac{[\text{CH}_3\text{COOH}]}{K_{\text{CA, Eriophorum}} + [\text{CH}_3\text{COOH}]} \frac{[\text{ATEA}]}{K_{\text{TE}} + [\text{ATEA}]} \]  

(E7)

where, \( V_{\text{CA, Eriophorum, Max}} \) \((10^{-5} \text{ mol electron kg}^{-1} \text{ SDW} \text{ (soil dry weight)} \text{ s}^{-1})\) at a reference temperature of 15°C is the maximum rate of consumption of acetate due to the nonmethanogenesis pathways (CA, Eriophorum), [CH\(_3\)COOH] and [ATEA] are concentrations of acetate (mol m\(^{-3}\)) and alternative terminal electron acceptors (ATEA; O\(_2\) under aerobic conditions and humic substances under anaerobic conditions in this study; mol electron m\(^{-3}\)), respectively, \( K_{\text{CA, Eriophorum}} \) and \( K_{\text{TE}} \) (Table 1) are the corresponding Michaelis constants, and 0.125 is the stoichiometric constant in electron acceptor reduction (i.e., 0.125 mol acetate needed per 1.0 mol reduced electron acceptor). Similar to the R1 to R3, \( V_{\text{CA, Sphagnum, Max}} \) is sensitive to temperature, and we use a \( Q_{10} \) value of 2.0 to account for the influences of temperature on the reaction rate.

Because DNDC does not explicitly simulate dynamics of organic electron acceptors provided by humic substances, their concentration (mol electron m\(^{-3}\)) is assumed to be linearly proportional to the concentration of DOC (i.e., equals to \( a \cdot (\text{DOC}) \)) in peat soils [Heitmann et al., 2007]. Acetate consumption due to oxidation by O\(_2\) occurs under aerobic conditions (e.g., in the unsaturated zone above the water table). We also use E7 to calculate the rate of this reaction, and set the \( \frac{[\text{ATEA}]}{K_{\text{TE}} + [\text{ATEA}]} \) term in E7 to 1.0, under the assumption that O\(_2\) is not a limiting factor for this process under aerobic conditions.

In the processes of acetate oxidation by O\(_2\) (under aerobic conditions) and ETAH (under anaerobic conditions), acetate is consumed and CO\(_2\) is produced. Changes in acetate and CO\(_2\) concentration are calculated as:

\[ V_{\text{CA, Eriophorum}} = -\frac{d[\text{CH}_3\text{COOH}]}{dt} = 0.5 \cdot \frac{d[\text{CO}_2]}{dt} \]  

(E8)

With these processes, DNDC explicitly simulates acetate dynamics as well as the pathways of acetotrophic and hydrogenotrophic methanogenesis by using acetate and H\(_2/\text{CO}_2\) as substrates. With this...
representation, acetate consumption due to the oxidation by O₂ and ETAH influences the acetate concentration in soils and therefore also influences CH₄ production under anaerobic conditions.

4.2. Modeling Stable C Isotopic Dynamics

To enable DNDC to predict dynamics of C isotopes, i.e., ^13C and ^12C transfers in the processes of CH₄ production, oxidation, and transport, we have converted each of the soil C pools into two new pools (i.e., ^13C and ^12C components; Figure 3) based on the principle of mass balance (E9) and δ^{13}C values (E10) for the soil C pools:

\[ M_C = M_{13C} + M_{12C} \]  
\[ \delta^{13}C = \frac{M_{13C}}{M_{12C}} - 1 \]  

where, \( M_C, M_{13C}, \) and \( M_{12C} \) are masses of C, ^13C, and ^12C in a soil C pool, respectively; \( \delta^{13}C \) is the C isotopic signature for the corresponding soil C pool and is expressed in parts per thousand (‰), VPDB is the standard δ^{13}C value (i.e., \((^{13}C/^{12}C)_{VPDB} = 0.0112372\) based on the ^13C and ^12C abundances in the Vienna Pee Dee Belemnite international standard material.

By solving the E9 and E10, the masses of ^13C and ^12C in a soil C-reactant pool (e.g., DOC) can be determined by using the mass of C and δ^{13}C value (a fixed value for initial carbon reactants, including DOC and root respired CO₂) as follows:

\[ M_{12C} = \frac{M_C \cdot (VPDB \cdot (1 + \delta^{13}C) + 1)}{1} \]  
\[ M_{13C} = M_C - M_{12C} \]  

Figure 3. The framework for simulating stable carbon isotopic dynamics in CH₄ transformations in the modified DNDC. Soil carbon pools have been converted into two new pools (i.e., ^13C and ^12C components) based on the principle of mass balance and δ^{13}C values for the soil C pools. DNDC calculates the ^13C and ^12C components of products by using the ^13C and ^12C components of reactants, simulated reaction rate, and value of isotopic fractionation factor, \( \alpha \), for an individual process in methane transformations. The model can predict δ^{13}C in soil CO₂ and CH₄ pools and emitted gas fluxes, which can be compared against observations from field studies or laboratory experiments for further testing and constraining simulations of the individual processes in CH₄ biogeochemistry. The abbreviations D, AM, HM, and MO indicate the processes of DOC decomposition, acetotrophic methanogenesis, hydrogenotrophic methanogenesis, and methane oxidation, respectively. Fluxes of CH₄ (FCH₄) are transported through plant (TP), ebullition (TE), and diffusion (TD). The processes of acetate consumptions due to the O₂ reduction and electron transfer between acetate and humic substances are not shown for the reason of clarity, and the ^13C and ^12C components of soil acetate and CO₂ pools are not changed by these processes because the fractionation factor, \( \alpha \), for them are 1.0 [Conrad, 2005; Corbett et al., 2013].
Isotopic fractionation in biogeochemical reactions (e.g., CH₄ production and oxidation) can be quantified by using a fractionation factor, \( \alpha \), defined as [Hayes, 1993]:

\[
\alpha = \frac{\left( ^{13}\text{C}/^{12}\text{C} \right)_{\text{reactant}}}{\left( ^{13}\text{C}/^{12}\text{C} \right)_{\text{product}}} \tag{E13}
\]

The value of \( \alpha \) can be determined by measuring the abundances of \( ^{13}\text{C} \) and \( ^{12}\text{C} \) in laboratory experiments. The sum of the produced \( ^{13}\text{C} \) and \( ^{12}\text{C} \) components equals the total C product based on the mass balance principle:

\[
P_C = P_{^{13}\text{C}} + P_{^{12}\text{C}} \tag{E14}
\]

where, \( P_C \), \( P_{^{13}\text{C}} \), and \( P_{^{12}\text{C}} \) are masses of C, \( ^{13}\text{C} \), and \( ^{12}\text{C} \), respectively, of total produced C in a C-producing reaction (e.g., the produced \( ^{13}\text{C} \)-CH₃COOH equals to the sum of the produced \( ^{13}\text{C} \)-CH₃COOH and \( ^{12}\text{C} \)-CH₃COOH for the R1).

By solving E13 and E14, the masses of \( ^{13}\text{C} \) and \( ^{12}\text{C} \) in a C product (e.g., CH₃COOH) can be calculated by using the \( ^{13}\text{C} \) and \( ^{12}\text{C} \) contents in the corresponding C precursors (e.g., DOC for the acetate production), fractionation factor (\( \alpha \)), and rate of the corresponding reaction (a variable simulated by DNDC) as follows:

\[
P_{^{13}\text{C}} = \frac{P_C \left( ^{13}\text{C}/^{12}\text{C} \right)_{\text{reactant}} / \left( ^{13}\text{C}/^{12}\text{C} \right)_{\text{product}}}{\alpha + 1} \tag{E15}
\]

\[
P_{^{12}\text{C}} = P_C - P_{^{13}\text{C}} \tag{E16}
\]

The principles of mass balance and C isotopic fractionation are applied to all processes in CH₄ cycling (Figure 3), although the fractionation factor (\( \alpha \)) may be close or equal to 1.0 (i.e., no isotopic fractionation effect) for some processes (e.g., anaerobic DOC decomposition and acetate consumptions due to \( \text{O}_2 \) reduction and \( \text{ETAH} \) pathways) [Conrad, 2005; Corbett et al., 2013]. For each process, the \( ^{13}\text{C} \) and \( ^{12}\text{C} \) components of products can be calculated by using the \( ^{13}\text{C} \) and \( ^{12}\text{C} \) components of reactants, reaction rate (simulated by DNDC), and isotopic fractionation factor (\( \alpha \)). To predict the \( ^{13}\text{C} \) of CH₄ fluxes, DNDC calculates: (1) the \( ^{13}\text{C} \) and \( ^{12}\text{C} \) components in DOC pool, (2) the \( ^{13}\text{C} \) and \( ^{12}\text{C} \) components in pools of C substrates (i.e., acetate and \( \text{CO}_2 \)) for CH₄ production, (3) production of acetate-induced \( ^{13}\text{CH}_4 \) and \( ^{12}\text{CH}_4 \), (4) production of \( \text{H}_2/\text{CO}_2 \)-induced \( ^{13}\text{CH}_4 \) and \( ^{12}\text{CH}_4 \), (5) \( ^{13}\text{C} \) and \( ^{12}\text{C} \) components in soil CH₄ and \( \text{CO}_2 \) pools by considering isotopic fractionation in the processes of CH₄ production and consumption, (6) \( ^{13}\text{CH}_4 \) and \( ^{12}\text{CH}_4 \) fluxes through each transport pathway (i.e., plant-mediated transport, ebullition, and diffusion) by considering isotopic fractionation in the processes of CH₄ transport, (7) \( ^{13}\text{C} \) of emitted gas fluxes, and (8) \( ^{13}\text{C} \) and \( ^{12}\text{C} \) components in remaining dissolved CH₄ and \( \text{CO}_2 \) pools (Figure 3). It should be noted that DNDC simulates CH₄ production after depletions of \( \text{NO}_3 \), \( \text{Mn}^{4+} \), \( \text{Fe}^{3+} \), and \( \text{SO}_4^{2-} \) and assumes that these nonmethanogenic reductions do not affect the \( ^{13}\text{C} \) value of the acetate for CH₄ production.

After incorporating the stable C isotopic dynamics into the DNDC’s framework, the model can predict \( ^{13}\text{C} \) in soil \( \text{CO}_2 \) and CH₄ pools and emitted gas flux. These outputs can be compared against observations from field studies or laboratory experiments for further testing and constraining simulations of the individual processes in CH₄ cycling.

5. Model Evaluation

We conducted model comparisons against observed data sets for two sites at Stordalen Mire, a moss-dominated site with degrading permafrost (active layer of about 1.2 m), and a wetter sedge-dominated site with no detectable underlying permafrost [McCalley et al., 2014]. The sites are referred to as Sphagnum and Eriophorum based on their dominant vegetation cover. Simulations of CH₄ fluxes, the relative contributions of the two CH₄ production pathways to the total production inferred from relative abundance of different methanogen functional groups, and C isotopic composition of CH₄ fluxes were compared to field data (see section 2). Simulations were run for both sites from 2010 to 2013, using daily meteorological data (i.e., maximum, mean, and minimum air temperature, precipitation, solar radiation, wind speed, and humidity) recorded at ANS (Figure 1). Simulated soil climate conditions were initialized by repeating the climate data in 2010 until the simulated annual mean soil temperature was stable. Then the vegetation and soil
biogeochemical modules were activated and the model was run for 2011 (calibration) and continuously from 2011 to 2013, with 2012 and 2013 for validation. Most input parameters (soils, hydrology, and vegetation) were set to the values used in previous simulations performed for the same sites [Deng et al., 2014]. In order to reduce the influence of WTD prediction-error on soil thermal and biogeochemical processes, we used the observed WTDs to drive the model if the measurements were available; if daily WTD data were not available, we interpolated daily values between observations as in Deng et al. [2014], using the hydrological parameters developed in that study. Based on new data from the Stordalen sites, pH (H2O) was set to 4.2 and 5.7, and peat C/N ratios were set to 60 and 30 for the Sphagnum and Eriophorum sites, respectively [Hodgkins et al., 2014].

Several new model parameters are needed to simulate acetate dynamics, methanogenesis, and stable carbon isotope fractionation (see equations (1), (3), (4), (7), (15), and (16); Figure 3). Parameters were set either to literature values or calibrated against field observations of CH4 fluxes and δ13C-CH4 in 2011 (Table 1). The calibrated parameters included maximum rate (V_{Max, DOC}) and half-saturation constant (K_{AnDecom}) for anaerobic DOC decomposition, maximum rates for AM (V_{AM, Max}) and HM (V_{HM, Max}), the coefficient quantifying the concentration of the electron acceptors provided by humic substances (a), and fractionation factors for AM (z_{AM}), HM (z_{HM}), CH4 oxidation(z_{MO}), and plant-mediated CH4 transport (z_{TP})V_{Max, DOC} and K_{AnDecom} were estimated by calibrating the simulated CH4 fluxes with the observed CH4 fluxes in 2011. The ratio of V_{AM, Max}:V_{HM, Max} was fixed by referring typical relative abundance of methanogenic groups (i.e., acetotrophic groups:hydrogenotrophic groups) for different ecosystems (0.1:0.9 for Sphagnum and 0.5:0.5 for Eriophorum) [Bridgham et al., 2013]. Maximum rates for AM and HM were then estimated by calibrating maximum rates of total CH4 production with the observed CH4 fluxes in 2011 and then distributing the calibrated value (8.0 mmol kg\(^{-1}\) dry peat d\(^{-1}\)) into AM and HM using the ratios of V_{AM, Max}:V_{HM, Max}. The coefficient, a, was estimated as 2.0 and 1.5 for the Sphagnum and Eriophorum sites, respectively, by constraining the parameter against the observed CH4/CO2 production in the field pore-water. The simulated production ratios of CH4/CO2 at the Sphagnum and Eriophorum sites were 0.19 and 0.40, respectively, in 2011, and were close to the observed CH4/CO2 production in the field pore-water sampled in June 2011 [Hodgkins et al., 2015]. The fractionation factors z_{AM}, z_{HM}, z_{MO}, and z_{TP} were estimated by calibrating the simulated δ13C-CH4 with the observed δ13C-CH4 in 2011 after fixing V_{Max, DOC}, K_{AnDecom}, V_{AM, Max}, V_{HM, Max}, and a. The reported ranges of z_{AM}, z_{HM}, z_{MO}, and z_{TP} were 1.000–1.032, 1.045–1.082, 1.007–1.031, and 1.012–1.021, respectively [Chanton et al., 1997; Conrad, 2005]. Through calibration with the δ13C-CH4 observations in 2011, the values of z_{AM}, z_{HM}, z_{MO}, and z_{TP} were set as 1.026, 1.073, 1.025, and 1.016, respectively. In general, the input parameters described above were primarily determined by calibrating against the field data of CH4 fluxes and δ13C-CH4 in 2011, and the calibrated model was then validated by comparing simulations against observations of CH4 fluxes and δ13C-CH4 in 2012 and 2013, as well as estimates of the relative contribution of the two CH4 production pathways.

5.1. Methane Fluxes
Simulated seasonal patterns of CH4 fluxes were close to the observed fluxes for both the calibration (2011) and validation (2012 and 2013) periods at the Sphagnum site. High peaks were noted in summer seasons from July to September in both the simulations and field records (Figure 4). In addition, DNDC simulated fluctuations of CH4 fluxes resulting from water table fluctuations (between −25 and 0 cm) at this site, consistent with observations (Figures 4a–4c). Correlation (R) values of model to field data were 0.86 and 0.61 in 2012 and 2013, respectively, indicating that the simulated seasonal variation of daily CH4 fluxes was significantly correlated with the corresponding observations in each year (p < 0.001). The simulated cumulative CH4 fluxes in 2012 and 2013 were 2.43 and 2.89 g CH4-C m\(^{-2}\), respectively, comparable with the corresponding observations of 2.91 and 2.45 g CH4-C m\(^{-2}\) with the root means square errors (RMSE) calculated as 0.48 and 0.44 g CH4-C m\(^{-2}\), respectively, in 2012 and 2013 (Table 2). The relative root means square errors (RRMSE) between the simulated and observed cumulative CH4 fluxes in 2012 and 2013 were 16% and 18%, respectively, which were less than the standard deviations of the observations in each year, meaning that the discrepancies between the model and field data were within the natural variation of the field observations.

At the Eriophorum site, the simulated seasonal patterns of CH4 fluxes by the modified DNDC were comparable with the observations (Figures 4d–4f); with R values of 0.86 and 0.68 in 2012 and 2013 (p < 0.001), respectively. However, simulated fluxes were higher than observations during the early growing season in
Daily CH$_4$ fluxes at the Eriophorum site were generally higher than that at Sphagnum. The observed cumulative CH$_4$ fluxes were 15.71 and 19.57 g CH$_4$-C m$^{-2}$ in 2012 and 2013, respectively, while the corresponding simulations were 14.38 and 22.72 g CH$_4$-C m$^{-2}$. The RMSE values were 1.33 and 3.15 g CH$_4$-C m$^{-2}$, respectively, for 2012 and 2013, and the corresponding RRMSE values were 8% and 16%, respectively (Table 2). Discrepancies between the simulations and observations were close to or less than the standard deviations of the observed cumulative CH$_4$ fluxes.

5.2. CH$_4$ Production Pathways

Simulated seasonal patterns for the HM and AM were similar in each year at Sphagnum, with relative high rates in summer (July–September) and reductions in the production rates following drops in the water table (Figures 5a–5c). HM was the primary CH$_4$ production pathway at the Sphagnum site, contributing

![Figure 4](image)

Figure 4. Water table dynamics (positive values for above-ground and negative values for below-ground), simulated, and observed daily CH$_4$ fluxes at the (a–c) Sphagnum and (d–f) Eriophorum sites during 2011–2013 growing seasons. Note the difference in CH$_4$ flux and water table scales between (left) Sphagnum and (right) Eriophorum plots. The correlations between the simulated and observed daily CH$_4$ fluxes were significant for all cases ($p < 0.001$). The observed CH$_4$ fluxes are the means of three (Sphagnum) or two (Eriophorum) chamber replicates and vertical bars are standard deviations of the replicate daily means.

<table>
<thead>
<tr>
<th>Year</th>
<th>Sphagnum</th>
<th>Eriophorum</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>O$^b$</td>
<td>M</td>
</tr>
<tr>
<td>2011</td>
<td>2.87(1.27)</td>
<td>2.00</td>
</tr>
<tr>
<td>2012</td>
<td>2.91(0.93)</td>
<td>2.43</td>
</tr>
<tr>
<td>2013</td>
<td>2.45(0.69)</td>
<td>2.89</td>
</tr>
</tbody>
</table>

$^a$The study period is the span during which continuous measurements of daily CH$_4$ fluxes were available. To calculate the total CH$_4$ emissions, over the sampling period in each year, fluxes for the days lacking measurements were determined using the arithmetic mean fluxes of the two closest days when observations were performed. Daily fluxes from either direct measurements or gap-filling were then summed up to calculate the growing period cumulative CH$_4$ emissions.

$^b$Each figure number within the bracket is the standard deviation of three (Sphagnum) or two (Eriophorum) replicate auto chamber plots.

$^c$RMSE and RRMSE are root mean squared error (g CH$_4$-C m$^{-2}$) and relative root mean squared error (%), respectively.
85% on average to the production (range = 36%–97%) over the three seasons from 2011 to 2013. We compared the simulated relative contribution of methanogenic pathways to total CH4 production against the observations of relative abundance of methanogen functional groups on individual days in 2011 (Table 3). The 10 day averages of the simulated contribution for hydrogenotrophic pathway through the growing season were 90%, 92%, 91%, and 82%, respectively, following 15 June, 12 July, 15 August, and 15 October. The simulations of the modified DNDC improved upon the results predicted by an early version without the aforementioned modifications [Deng et al., 2014], and were comparable with the corresponding observations of 90%, 85%, 94%, and 85%, respectively (Table 3; note that the AM fraction equals 1.0 minus the HM fraction).

At the Eriophorum site, simulated high peaks of methanogenesis occurred from July to September, with the AM and HM pathways contributed approximately equally to total methanogenesis during the growing season (Figures 5d–5f). The average contributions of hydrogenotrophic and acetotrophic pathways to total CH4 production were calculated as 51% (range: 35%–61%) and 49% (range: 39%–65%), respectively, across

Figure 5. Simulated daily CH4 production (brown line; mg CH4-C m^-2 d^-1) at the (a–c) Sphagnum and (d–f) Eriophorum sites during 2011–2013, and its partitioning into hydrogenotrophic methanogenesis (HM, green shading) and acetotrophic methanogenesis (AM, blue diagonal lines), respectively. The grey lines represent the fractional contribution of HM to total CH4 production.

Table 3. Modeled (M) and Observed (O) Relative Fractional Contribution of the Hydrogenotrophic Methanogenic Pathway (HM) to Total CH4 Production at the Stordalen Sphagnum and Eriophorum Sites on Individual Days in 2011

<table>
<thead>
<tr>
<th>Date</th>
<th>Sphagnum</th>
<th>Eriophorum</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>M_h</td>
<td>M_o</td>
</tr>
<tr>
<td>15 Jun</td>
<td>0.44</td>
<td>0.45</td>
</tr>
<tr>
<td>12 Jul</td>
<td>0.46</td>
<td>0.49</td>
</tr>
<tr>
<td>16 Aug</td>
<td>0.40</td>
<td>0.59</td>
</tr>
<tr>
<td>16 Oct</td>
<td>0.59</td>
<td>0.51</td>
</tr>
</tbody>
</table>

The acetoclastic methanogenic pathway (AM) fraction equals 1.0 minus this fraction.

Observed production pathway partitioning was inferred from the relative abundance of methanogen functional groups [McCalley et al., 2014].

Modeled results are 10 day averages of contribution of hydrogenotrophic and acetotrophic pathways following the corresponding date with observation, and were produced with the modified (M_h) and an older (M_o) version of DNDC. The old version in Deng et al. [2014] did not explicitly simulate acetate dynamics and used DOC as a substrate for acetoclastic methanogenesis.
the three seasons from 2011 to 2013. The 10 days averages of the simulated contributions for HM were 53%, 49%, 49%, and 58%, respectively, following 15 June, 12 July, 15 August, and 15 October in 2011. The simulations by the modified DNDC improved upon the results from an early version without modifications, and were comparable with the corresponding observations, which were 68%, 61%, 59%, and 54%, respectively (Table 3; note that the AM fraction equals 1.0 minus the HM fraction).

5.3. CH₄ Flux Isotopic Composition

Unlike the flux rates, no clear seasonal pattern appeared for δ¹³C-CH₄ at either the Sphagnum or Eriophorum sites. And there is no correlation between the δ¹³C-CH₄ and flux rates. However, both the simulations and observations showed a consistent divergence of around 8% between these two sites during both the calibration (2011) and validation (2012 and 2013) periods although the divergence was occasionally overlain by fluctuations of δ¹³C-CH₄, likely due to seasonal fluctuations in the rates of CH₄ production, oxidation, and transport (Figure 6).

The means of the observed δ¹³C-CH₄ at the Sphagnum and Eriophorum sites were −79‰ and −67‰, respectively, for the calibration year 2011. For the validation periods in 2012 and 2013, the means of the observed δ¹³C-CH₄ at Sphagnum and Eriophorum were −71‰ and −65‰, respectively. The DNDC simulated corresponding δ¹³C-CH₄ at the Sphagnum and Eriophorum sites were −76‰ and −67‰ (2011) and −71‰ and −66‰ (2012 and 2013), respectively. Both the simulations and observations showed that most of the δ¹³C-CH₄ values distributed below −65‰ for Sphagnum and between −70‰ and −60‰ for Eriophorum (Figure 7). The simulations by the modified DNDC improved upon the predictions by the version without acetate dynamics, which failed to capture the observed relative contribution of methanogenic pathways to total CH₄ production (Table 3) and thereafter predicted δ¹³C-CH₄ with systematic discrepancies in comparison with the field records (Table 4).

Figure 6. Simulated and observed δ¹³C of daily CH₄ fluxes at the Sphagnum and Eriophorum sites during 2011–2013. The observed δ¹³C values are the means of three (Sphagnum) or two (Eriophorum) chamber replicates and vertical bars are standard deviations of the replicates (McCally et al., 2014).
6. Sensitivity Analysis

To investigate the general behaviors of the modified DNDC in predicting C isotopic signature in CH4 fluxes, we conducted a sensitivity analysis by varying several input parameters that have an uncertainty range, including the C isotopic signatures for DOC ($\delta^{13}$C-DOC) and root respired CO2 ($\delta^{13}$C-CO2-root), and fractionation factors for the processes of CH4 production ($\alpha_{AM}$ and $\alpha_{HM}$), CH4 oxidation ($\alpha_{MO}$), and plant-mediated CH4 transport ($\alpha_{TP}$). The baseline scenario was set based on the actual conditions at Eriophorum, and alternative scenarios were created by varying a single input parameter while keeping others constant. The varied ranges were $2^{30}$ to $2^{22}$ for $\delta^{13}$C-DOC and $\delta^{13}$C-CO2-root, 1.000–1.032 for $\alpha_{AM}$, 1.045–1.082 for $\alpha_{HM}$, 1.007–1.031 for $\alpha_{MO}$, and 1.012–1.021 for $\alpha_{TP}$ [Chanton et al., 1997; Conrad, 2005; Ehleringer et al., 2000]. Mean simulated $\delta^{13}$C-CH4 from 1 June to 31 October 2013 under each of the scenarios were compared with a sensitivity index (SI) [Nearing et al., 1990] calculated as follows:

$$SI = ((O_2 - O_1)/O_{avg})/((I_2 - I_1)/I_{avg}) \quad (E17)$$

where $I_1$, $I_2$, and $I_{avg}$ are the minimum, maximum, and average values of a selected input parameter, $O_1$, $O_2$, and $O_{avg}$ are the corresponding modeled $\delta^{13}$C-CH4.

The sensitivity results (Table 5) indicate that (1) $\delta^{13}$C in emitted CH4 was very sensitive to variation in $\alpha_{AM}$, $\alpha_{MO}$, and $\alpha_{TP}$ with more enriched $\delta^{13}$C resulting from smaller fractionation effects in CH4 production ($\alpha_{AM}$ and $\alpha_{MO}$) and larger fractionation effects in CH4 transport ($\alpha_{TP}$) exerted a moderate influence on $\delta^{13}$C-CH4 as compared to other factors, and larger fractionation effects in CH4 transport resulted in smaller $\delta^{13}$C-CH4 (i.e., more depleted $^{13}$C) for the test case.

7. Discussion

Considering microbial communities when developing ecosystem models is an important next step toward accurately simulating ecosystem processes and predicting responses to environmental change [Wieder et al., 2013; Graham et al., 2012, 2014]. For example, recent research at Stordalen Mire has highlighted the need to consider microbial ecology and its impact on CH4 production and isotopes when modeling the CH4 emissions from thawing permafrost [McCalley et al., 2014]. The challenge is how to model shifts in microbial...
metabolism and the resulting change in CH₄ emissions and δ¹³C-CH₄ across spatially complex, thawing northern peatlands. Incorporating stable isotopic dynamics into traditional biogeochemical models is at an early stage. There are no other CH₄ models that can simulate stable carbon isotopic dynamics [Xu et al., 2016].

In this study, we used microbial and isotopic field data to identify a shortcoming in DNDC simulations of CH₄ production pathways in wetland sites. The initial disconnect between DNDC outputs and field measurements of CH₄ production pathway, despite reasonable CH₄ flux simulations [Deng et al., 2014] and incorporation of both AM and HM pathways [Fumoto et al., 2008] was primarily due to using bulk DOC as a substrate for acetoclastic methanogenesis. The disconnect exemplifies how models can get the correct net CH₄ emission, while missing or misrepresenting key underlying mechanisms. Consideration of isotopic and microbial prevalence data led to the addition of an explicit acetate metabolism pathway providing the substrate for acetotrophic methanogenesis, which appreciably improved DNDC’s simulation of the relative contributions of AM and HM pathways to total CH₄ emission (Figure 8). Further, incorporation of stable C isotope dynamics into DNDC now provides an additional internal metric for testing and constraining the processes (production, oxidation, transport) that underlie CH₄ fluxes in DNDC. We note that adding the new processes also introduced new parameters that are not well-enough constrained by literature values and required calibration.

We also modified DNDC by explicitly simulating acetate dynamics and pathways of acetotrophic and hydrogenotrophic methanogenesis, which we hypothesized could improve DNDC’s simulation of the relative contribution of methanogenic pathways to total CH₄ production. Model tests against field measurements indicated that the modified DNDC successfully captured the differences between the Sphagnum and Eriophorum sites in total CH₄ fluxes, relative contribution of methanogenic pathways to CH₄ production, and the δ¹³C isotopic signature of emitted CH₄ (Figure 8). In earlier analyses, we found that CH₄ flux differences between the Sphagnum and Eriophorum sites were a result of site differences in soil environments and vegetation characteristics [Deng et al., 2014].

Table 5. Calculated Sensitivity Indices Quantifying the Impacts of Variations in C Isotopic Signature of Substrates (DOC and Root Respired CO₂) and Fractionation Factors a on δ¹³C-CH₄ Predicted by the Modified DNDC

<table>
<thead>
<tr>
<th>Items</th>
<th>δ¹³C-DOC</th>
<th>δ¹³C-CO₂</th>
<th>a</th>
<th>b</th>
<th>c</th>
<th>d</th>
</tr>
</thead>
<tbody>
<tr>
<td>Average</td>
<td>-26°</td>
<td>-26°</td>
<td>1.016</td>
<td>1.064</td>
<td>1.019</td>
<td>1.017</td>
</tr>
<tr>
<td>Minimum</td>
<td>-30°</td>
<td>-30°</td>
<td>1.000</td>
<td>1.045</td>
<td>1.007</td>
<td>1.012</td>
</tr>
<tr>
<td>Maximum</td>
<td>-22°</td>
<td>-22°</td>
<td>1.032</td>
<td>1.082</td>
<td>1.031</td>
<td>1.021</td>
</tr>
<tr>
<td>SIB</td>
<td>0.21</td>
<td>0.15</td>
<td>8.82</td>
<td>6.85</td>
<td>-5.58</td>
<td>2.82</td>
</tr>
</tbody>
</table>

a, b, c, and d are fractionation factors for the processes of acetotrophic methanogenesis, hydrogenotrophic methanogenesis, CH₄ oxidation, and plant-mediated CH₄ transport, respectively. The uncertainty ranges of δ¹³C-DOC, δ¹³C-CO₂, a, b, c, and d were determined from Chanton et al. [1997], Conrad [2005], and Ehleringer et al. [2000].

bSI is a relative sensitivity index, the higher the absolute value of the SI, the greater the impact the input has on the output.

Figure 8. Mean CH₄ flux (bars with interannual standard deviation, left axis) and δ¹³C of emitted methane (diamonds, right axis) during three study periods from 2011 to 2013, and 2011 relative contribution of hydrogenotrophic and acetotrophic methane production pathways (pie charts, observed inferred from the relative abundance of methanogen functional groups) [McCalley et al., 2014] at the Eriophorum and Sphagnum sites. Comparison across an older DNDC (DNDCₒ) version [Deng et al., 2014], the modified DNDC (DNDCₘ), and field observations at Stordalen mire in Sweden.
contribution of methanogenic pathways could be primarily attributed to different vegetation characteristics between these two sites. Based on E3 and E4, the relative contribution of methanogenic pathways to total CH$_4$ production is jointly controlled by the ratio of $V_{AM,\,Max}$: $V_{HM,\,Max}$ and substrate concentration, which were either based on vegetation types (Table 1) or simulated by tracking substrate production and consumption closely related to vegetation characteristics. The differences in CH$_4$ metabolism between the Sphagnum and Eriophorum sites are representative of divergences in CH$_4$ metabolism between bog and fen habitats generally [Alstad and Whiticar, 2011; Chanton et al., 2003; Galand et al., 2010], suggesting that our modifications to DNDC should improve its ability to accurately simulate methanogen community dynamics and $\delta^{13}$C signatures of emitted CH$_4$ across northern peatlands. Acetate dynamics played an important role in predicting the contribution of CH$_4$ production pathways. For example, without acetate consumption by alternative (nonmethanogenic) processes (Figure 2), the simulated relative contribution of AM and HM to total CH$_4$ production at the Sphagnum site was about 2:1, due to substrates’ (acetate and H$_2$) availability determined by stoichiometry in CH$_4$ production processes (R1 to R3), even though $V_{AM,\,Max}$:$V_{HM,\,Max}$ was set as 1:9, based on prior knowledge of relative abundance of methanogenic groups in this ecosystem [Bridgham et al., 2013]. Acetate dynamics will also influence total CH$_4$ emissions in the current DNDC framework because acetate availability for CH$_4$ production is reduced by consumption by alternative reduction processes. Therefore, explicitly simulating acetate dynamics may be necessary for reliably simulating the contribution of AM and HM to total CH$_4$ production and CH$_4$ emissions from northern peatlands.

We note some discrepancies between the modeled results and field measurements. DNDC overestimates CH$_4$ emission rates during the early growing season (mid-June to mid-July) in 2013 at both the Sphagnum and Eriophorum sites, which may have resulted from either overpredictions of substrate availability (i.e., acetate and H$_2$) due to relative high air temperature during this period (mean: 11.2°C from 15 June to 15 July), or oversensitivity of CH$_4$ production to temperature ($Q_{10} = 4.6$ for CH$_4$ production) [Van Bodegom and Scholten, 2001], causing DNDC to predict relatively higher CH$_4$ production and emissions. Because meteorological data at ANS (10 km northwest of Sordalen) were used to support the simulations, and soil climate, plant growth, and CH$_4$ transformations are strongly related to temperature, deviations in predicting daily variability in CH$_4$ fluxes may in part be caused by a lack of site-specific meteorological data.

Discrepancies also exist between the modeled and observed daily $\delta^{13}$C-CH$_4$ fluxes. For example, the model predicted more depleted $^{13}$C in emitted CH$_4$ (i.e., more negative $\delta^{13}$C-CH$_4$) during early July to mid-August in 2012 at the Sphagnum site than in 2011 or 2013, while measured $^{13}$C-CH$_4$ was less depleted in 2012 than in 2011 or 2013 (Figure 6). In the DNDC simulation, HM was a slightly larger fraction of total methanogenesis during early July to mid-August in 2012 than in 2011 or 2013 (Figure 5). DNDC results could be interpreted as a hypothesis that the relative abundance of active hydrogenotrophic methanogens would be slightly higher during early July to mid-August in 2012 than in 2011 or 2013 (Table 3 has data for 2011 only). Another possible explanation for the discrepancies between the modeled and measured $\delta^{13}$C-CH$_4$ at the Sphagnum site could be discrepancies in simulations of CH$_4$ oxidation and/or $\delta^{13}$C in soil CO$_2$ pool, considering the negligible fractionation effects for CH$_4$ transport at this moss-dominated site. This could be evaluated by comparisons between the simulated and observed $\delta^{13}$C in the soil CO$_2$ pool.

Several new parameters were estimated by calibrating the simulations with the field observations in 2011 due to a lack of specific parameter values in the literature (Table 1). Most of the parameters estimated from the calibration were comparable with ranges in published data for similar ecosystem types. For example, the maximum rate of total CH$_4$ production (8.0 mmol kg$^{-1}$ dry peat d$^{-1}$) was within the range reported for pan-Arctic Graminoid [Treat et al., 2015]. However, little information can be found for the parameters $V_{Max,\,DOC}$, $K_{AndDecomp}$, and $a$, and this lacking of parameter information suggests that some important processes regarding acetate dynamics and methanogenesis need to be better quantified. More accurate parameterization may also be helpful for improving model predictions of CH$_4$ emission. For instance, the concentration of the electron acceptors provided by humic substances is assumed to be linearly proportional to DOC concentration, and the proportionality constant has been set to one value for each type of vegetation community by constraining this parameter against the observed pore-water CO$_2$:CH$_4$ ratio. This simplification could reflect a general characteristic of the test sites, but may be not enough to capture daily variability in the concentration of the alternative electron acceptors because the proportionality constant may actually vary across a growing season [e.g., Heitmann et al., 2007]. Since acetate consumption due to the ETAH pathway is a process directly competing with AM in the model, this
constant proportionality assumption may have a notable impact on the daily variability of the simulated CH₄ emissions, and needs further testing.

References


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